



POLYSUBSTITUTED 2H-PYRAN-2-ONES : A NEW CLASS OF HEPATOPROTECTIVE AGENTS⁺

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Abstract : The synthesis and hepatoprotective activity of 6-aryl-3-substituted-4-methylthio-2H-pyran-2-ones (**3**, **5**, **7**) and the corresponding 2-thiones (**6**), 6-aryl-4-methylthio-2H-pyran-2-ones (**4**) and 6-aryl-3-carbethoxy-5-cyano-4-methylthio-2H-pyran-2-ones are described. © 1997 Elsevier Science Ltd.

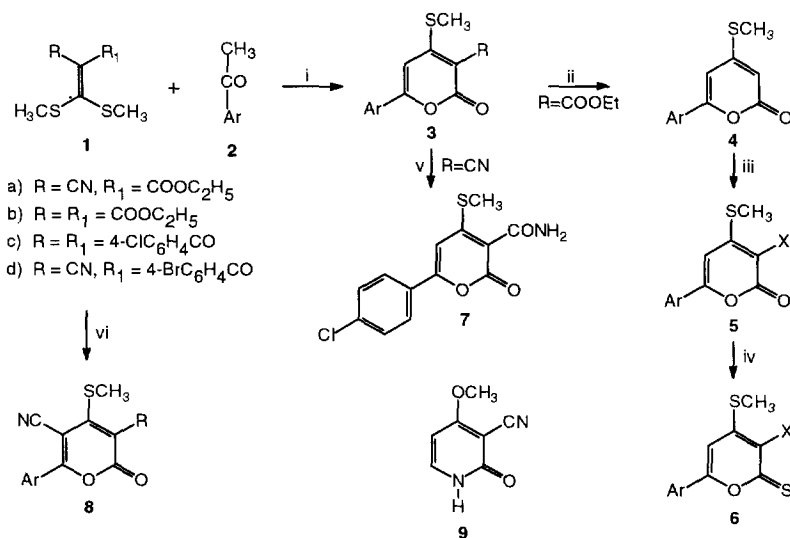
Perpetual exposure of liver to xenobiotics and therapeutic agents leading to toxic manifestations of complex and diverse nature, on one hand and paucity of any safe medication on the other, renders the management of hepatic disorders more problematic. The use of some synthetic and semisynthetic drugs which include steroids, immunosuppressive agents and antiviral drugs such as corticosteroids, acyclovir, vidrabin alone or in combination with α -interferon, has been a subject of controversy and met with limited survival rates. Besides the risk of relapse and side effects, these provide only symptomatic relief and have little influence on the disease process itself.

Many herbal formulations, whose efficacies are long known, yet little understood, are claimed to be effective in hepatic diseases. Several class of natural products¹, comprising the constituents of herbs in the formulation, various sulfur compounds²⁻⁴ display pronounced hepatoprotective activity and many of these possess pyran-2- or 4-one moieties either in a rigid or flexible form. Recently, 3-cyano-4-methoxy-1H-pyridin-2-one^{1b} (**9**), isolated from *Ricinus communis*, was found to be effective at 6 mg/kg/p.o. dose with LD₅₀ 80 mg/kg/p.o. in rats in various models of liver injuries. This provided a template for designing and eventual emergence of the present new class of hepatoprotectants^{5,6}. Accordingly, retaining some of the structural features of **9** on 2H-pyran-2-one or the corresponding thione ring skeletons, permitted in identifying a series of 6-aryl-3-carbethoxy-, 6-aryl-3-cyano-, and 6-aryl-3-carboxamido-4-methylthio-2H-pyran-2-ones (**3**, **7**), 6-aryl-3-halo-4-methylthio-2H-pyran-2-ones (**5**) and the corresponding thiones (**6**), 6-aryl-4-methylthio-2H-pyran-2-ones (**4**) and 6-aryl-3-carbethoxy-5-cyano-4-methylthio-2H-pyran-2-ones (**8**) [Scheme 1] to be effective in liver injuries. Among these, the 6-(4-chlorophenyl)-3-cyano-4-methylthio- and 3-bromo-6-(4-chlorophenyl)-4-methylthio-2H-pyran-2-ones (**3b**, **5d**) were most potent of the series.

All the synthesized compounds listed in Table 1 were screened for their hepatoprotective activity against thioacetamide-induced hepatic damage in rats. The activity of the compounds was assessed on the basis of % protection afforded in various levels of serum enzymes such as glutamate pyruvate transaminase

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Scheme 1



Reagents and conditions: i) DMF/KOH or $\text{K}_2\text{CO}_3/\text{DMSO}$, RT; ii) PPA, 120°C ; iii) NCS or NBS in CH_2Cl_2 ; iv) $\text{P}_2\text{S}_5/\text{Toluene}$; v) H_2SO_4 , 60°C ; vi) $\text{NCCH}_2\text{COOC}_2\text{H}_5/\text{DMSO}/\text{K}_2\text{CO}_3$

(GPT)⁷, glutamate oxaloacetate transaminase⁷(GOT), acid phosphatase⁸(ACP), alkaline phosphatase⁹(ALP) and glutamate dehydrogenase¹⁰(GIDH). The results are presented in Table 1. The compound **3b** was studied in detail in different toxin-induced models and % protection is reported in Table 2. A study of structural requirements for activity in compounds **3 - 8**, revealed that substituents at C-3 and C-6 positions in the pyran ring play a pivotal role in expressing significant hepatic protection as adjudged by bringing stepwise change in the ring substituents.

A change in the 6-aryl substituents in **3 - 8** rendered the order of activity as 4-chlorophenyl > 2,4-dichlorophenyl > 4-bromophenyl > 4-fluorophenyl. Switching over from aromatics to heteroaromatics, the activity of moderate order was retained only in **3e, f**, while compounds with pyridyl substituents (**3g, h**) were devoid of activity.

Among the various substituents at C-3, the CN (**3a,h**), Br (**5c-g, 6b**) and Cl (**5a, b, 6b**) were found to optimise the activity in compounds bearing haloaryl, 2-thienyl or 2-furyl substituents at C-6 position. Shifting the CN group from C-3 to C-5 gave inactive compound (**8**). Compounds **3b** and **5d** were almost equipotent in offering protection in all the enzyme parameters due to closeness in the atomic volumes of CN (23.5) and Br (22.3). However, **5d** had an edge over **3b** in displaying better protection (GPT, ALP). Compounds without substituents at C-3 (**4a, b**) exhibited moderate hepatoprotective activity. Replacement of CN by COOC_2H_5 (**3i**) and CONH_2 (**7**) groups led to the loss of activity. Thus, it may be concluded that 6-haloaryl with 3-cyano or 3-bromo substituents at 4-methylthio-2H-pyran-2-one nucleus seem to be essential requirements for expressing hepatoprotective activity.

Table 1 : Hepatoprotective activity of **3 - 8** against thioacetamide induced toxicity in rats at 6 mg/kg (p.o. x 7 days). Values are the % protection afforded by the compounds

Compd.	Ar	R/X	GOT	GPT	ACP	ALP	GIDH
3a	4-FC ₆ H ₄	CN	20	31	46*	15	0
3b	4-ClC ₆ H ₄	CN	68**	69**	59*	65*	51*
3c	4-BrC ₆ H ₄	CN	20	10	38	56*	40
3d	2,4-Cl ₂ C ₆ H ₃	CN	37	45	61*	33	65**
3e	2-Thienyl	CN	37	74**	10	33	100**
3f	2-Furyl	CN	71**	72**	0	19	15
3g	3-Pyridyl	CN	12	12	38	0	51*
3h	4-Pyridyl	CN	0	0	0	12	0
3i	4-ClC ₆ H ₄	COOC ₂ H ₅	23	31	0	11	17
4a	4-FC ₆ H ₄	H	59*	57*	46*	30	15
4b	4-ClC ₆ H ₄	H	46*	51	24	42	43
5a	4-ClC ₆ H ₄	Cl	36	51	38	40	27
5b	4-BrC ₆ H ₄	Cl	30	40	30	19	26
5c	4-FC ₆ H ₄	Br	0	0	49	43	0
5d	4-ClC ₆ H ₄	Br	70**	80*	48*	100**	51*
5e	4-BrC ₆ H ₄	Br	47*	18	72*	15	28
5f	2-Thienyl	Br	50*	55*	25	37	4
5g	4-Pyridyl	Br	48*	0	53*	12	16
6a	4-ClC ₆ H ₄	Cl	29	26	25	10	12
6b	4-ClC ₆ H ₄	Br	25	0	0	36	20
7	4-ClC ₆ H ₄	CONH ₂	16	15	10	17	0
8a	4-ClC ₆ H ₄	COOC ₂ H ₅	14	19	30	0	16
8b	4-BrC ₆ H ₄	COOC ₂ H ₅	0	0	2	0	2
	Silymarin (Standard drug)		50.14*	47.25*	43.25*	47*	38.81

**P < 0.01; *P < 0.05) as compared to toxin treated group.

Table 2 : Hepatoprotective activity of 6-(4-chlorophenyl)-3-cyano-4-methylthio-2H-pyran-2-one (**3b**) in different models at 6 mg/kg (p.o. x 7 days) is expressed in terms of % protection. Values in parentheses are due to Silymarin, used as a standard drug in identical conditions.

Models	GOT	GPT	ACP	ALP	GIDH
Galactosamine	86** (47)	76** (45)	66** (61)	68* (33)	57* (65)
Paracetamol	73** (46)*	81** (49)*	90** (67)*	73* (50)*	54* (60)*
Carbon Tetrachloride	59* (48)*	67* (64)*	70* (57)*	57* (51)*	58* (51)*
Alcohol	34 (43)	48 (59)*	53* (49)*	24 (62)*	40 (49)
Alcohol 12 mg/kg (p.o. x 7 days)	58* (62)*	69** (78)*	74** (64)**	56* (77)**	62* (59)*

**P < 0.01; *P < 0.05 > as compared to toxin treated group.

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$$\frac{(\text{Toxin treated}) - (\text{Toxin + test compound treated})}{\text{Normal} - (\text{Toxin treated})} \times 100$$

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